

Study of Biologically Active Substances in Gel Obtained on the Basis of Local Plant Raw Materials

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Abstract In this article, a quantitative analysis of biologically active substances in a gel obtained from local plant materials was carried out. The aerial part of plants (*Urtica dioica* U.), (*Polygonum hydropiper* L.) and (*Polygonum aviculare* L.) were used as local plant materials. Studies of literature data showed that these medicinal plants contain valuable biologically active substances that have a healing effect on the human body. The article deals with the development of gel technology based on the liquid extract "Hemostat". The choice of the base and excipients, the specifics of the technology. Special attention is paid to the choice of base concentration. Using technological methods of analysis, the composition and technology of a dental gel based on the liquid extract "Hemostat" FSP 42 Uz-22477731-3757-2019 was experimentally substantiated and developed. A promising and fundamental method for the treatment of diseases of the oral mucosa is the use of rational dosage forms - a transdermal system - a gel with good absorption based on active medicinal substances. Using high performance liquid chromatography, the qualitative and quantitative content of flavonoids and vitamin K₁ was studied.

Keywords Gel, Base, *Urtica dioica*, *Polygonum hydropiper*, *Polygonum aviculare*, High performance liquid chromatography, Quantitative determination of flavonoids, Rutin, Quercetin, Vitamin K₁

1. Introduction

Despite the rich range of drugs of synthetic origin, plant-based drugs do not lose their popularity. In dental practice, for the treatment of inflammation in the mouth and gums, more and more preference is given to the use of herbal medicine [1-8]. Phytotherapy is a method of treatment based on the use of medicinal plants and complex preparations based on them. Plants synthesize aromatic substances, most of which are phenols and their oxygen-substituting derivatives, such as tannins, useful for maintaining human and animal health. [8-15] Many of them, in particular alkaloids, are defense mechanisms of plants against microorganisms, insects and herbivores [16]. Therefore, phytopreparations, as a rule, have a pronounced immunostimulating effect (effect). Stinging *Urtica dioica* L., family Urticaceae-nettles is one of the most popular medicinal plants in traditional and folk medicine. Preparations from *Urtica dioica*s are used as a hemostatic, anti-inflammatory, antitumor, and choleric agent. [17-19] Medications based on stinging nettle "Prostaforton" and "Bazoton" are used to treat prostate adenoma and

prostatitis, which are among the most common diseases in men [19-25]. The beneficial properties of medicinal plants depend on the content of so-called active substances in them, such as macro- and microelements, vitamins K₁, B, flavonoids, saponins, alkaloids, essential oils, organic substances [26-30]. Infusion, liquid extract, powder and juice of *Urtica dioica* are used for acute and chronic enterocolitis with intestinal, uterine bleeding and as a multivitamin and diuretic [30-32].

The pharmacological action of drugs based on the herb of *Polygonum hydropiper* L., Buckwheat family - Polygonaceae is due to a number of substances, including flavonoids, vitamin K₁, terpenoids, carotenoids, however, literary data regarding their component composition are quite contradictory [1-3]. So, in the domestic literature it is reported that the herb of *Polygonum hydropiper* contains quercetin, kaempferol, luteolin, myricetin, isorhamnetin, rhamnasin, quercitrin, hyperoside, persicarin 7-methyl ether, rutin. According to foreign scientists, the herb of this plant, along with quercetin and quercitrin, contains taxifolin, quercetin-3-sulfate. [1-5]

The herb *Polygonum aviculare* L. is used as a source of flavonoids that have anti-inflammatory and vasoconstrictive effects. [8]. Infusions of herb *Polygonum aviculare* are effective in bleeding, sexual disorders. Knotweed has antibacterial and antiviral (anti-HIV) activity, is part of medicinal preparations (phytolysin, tussiflorin, Zdenko collection). Herbal preparations of *P. aviculare* are used for urolithiasis, uric acid diathesis,

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and skin diseases [12-15]. They are also effective in functional insufficiency of the liver and kidneys, inhibit the development of liver fibrosis. Glycosylated quercetin derivatives are used in cancer prevention [16-18].

The aim of the work is to study the qualitative and quantitative content of flavonoids, carotenoids, and vitamin K₁ in the gel obtained from local plant materials using high performance liquid chromatography [20].

2. Materials and Methods

To obtain the gel, we chose as an active pharmaceutical ingredient: liquid extract "Hemostat", with a pronounced anti-inflammatory and hemostatic effect, which helps wound healing and bleeding of the gums [9].

The following were used as excipients: sodium carboxymethyl cellulose (NaCMC - the basis of the gel), nipagin (preservative), purified water (solvent, pharmacologically indifferent, is a universal solvent for a number of biologically active substances), glycerin (plasticizer, is a viscous transparent, colorless liquid, unlimitedly soluble

in water. Sweet in taste, which is why it got its name, it dissolves many substances well). Hydrophilic pressure-sensitive bases can provide high adhesion to both dry substances and liquid extracts, and are also suitable for use in the development of the composition of dental gels [9]. To increase the plasticity of the gel, glycerol was introduced at a concentration of 3-15%. The results of the experiment are presented in Table 1.

Table 1. Characteristics of the gel with different concentrations of glycerin

Quality characteristics	The concentration of glycerin in the gel			
	5%	8%	10%	12%
Appearance	Does not match	Does not match	Corresponds	Does not match

As can be seen from the table, when adding different concentrations of 5-8% glycerol, it was not enough to create a plastic gel; when adding 12-15% glycerol, the base liquefied and a liquid mass was obtained [9]. Based on this, we can conclude that containing glycerol in a concentration of 10% provides good quality. The compositions of the studied compositions are presented in Table 2.

Table 2. Selection of the concentration of the gel base

№ p / p composition	Base Components				
	Na-CMC	Glycerol	Nipagin	Purified water, g	Basis evaluation
1	2,0	10,0	0,2	up to 100	Liquid consistency
2	4,0	10,0	0,2	up to 100	Appropriate consistency
3	10,0	10,0	0,2	up to 100	Thick mass
4	15,0	10,0	0,2	up to 100	The presence of lumps
5	20,0	10,0	0,2	up to 100	Highly viscous mass
6	25,0	10,0	0,2	up to 100	Does not match

The criteria for selecting the amount of the base were the following property indicators, mass homogeneity, gel quality uniformity: description, appearance, adhesive consistency. The base of the selected form is well spread on the oral mucosa, has a pleasant smell and texture, the rheological properties meet the requirements for soft dosage forms according to the SP XI edition and the State Pharmacopoeia of the Republic of Uzbekistan [9]. Transparent, homogeneous, viscous mass, easily distributed on the mucous membrane.

Table 3. The composition of the gel based on №. 2

№	Base components, Mr.		
1	liquid extract "Hemostat"	10,0	active substance
2	Na-CMC	4,0	gel base
3	glycerol	20,0	plasticizer
4	nipagin	0,2-0,5	preservative
5	purified water	up to 100,0	solvent

The data obtained show that the highest value of the generalized desirability function in terms of consistency,

mass homogeneity, appearance, dissolution time, adhesive properties, mechanical strength, has composition №. 2. The composition of the selected gel (№. 2 model) is presented in Table 3.

Important technological indicators of the gel include the structural and mechanical properties of the gel, which affect such therapeutic and consumer indicators as the release of medicinal substances, ease of use and ease of application.

Table 4. Gradient mode

Time, min	Mobile phase A, %	Mobile phase B, %
3,00	80,0	20,0
5,00	70,0	30,0
7,00	70,0	30,0
10,00	50,0	50,0
14,00	50,0	50,0
16,00	30,0	70,0
20,00	30,0	70,0
22,00	80,0	20,0
25,00	Finish	

In order to study the qualitative and quantitative content of the sum of flavonoids and vitamin K₁ contained in the gel, the HPLC method was used.

To determine flavonoids, chromatography is carried out on an Agilent Technologies (USA) liquid chromatograph, Agilent 1200 series, with Chemstation 09.03.a software, a gradient pump, and a spectrophotometric detector. Separation is carried out on a 4.6x150 mm column filled with Zorbax EclipseC-18 sorbent with a particle size of 5 μm. Mobile phase 0.3% phosphoric acid (A) and methanol (B), gradient mode are presented in Table-4 [14-15].

10 μl of the obtained extract was injected into the chromatograph and chromatographed at a rate of 1 ml/min, the volume of the injected sample was 10 μl. Detection was carried out at wavelengths of 370 nm, which are the characteristic L_{max} of rutin and quercetin, respectively. Chromatography temperature - 350C. The duration of the analysis was 30 min. In order to identify rutin and quercetin in the preparation, their solutions were chromatographed in parallel with RSO. The analysis was carried out in five repetitions. Retention times for rutin (11.72 min) and quercetin (14.28 min). The content of rutin and quercetin in the composition of the liquid extract was calculated using the following formula:

$$X = \frac{S_{test} \times a_{st} \times V_{test} \times P \times 1000}{S_{st} \times V_{st} \times a_{test} \times 100}$$

$$= \frac{S_{test} \times a_{st} \times P \div V_{test} \times 10}{S_{st} \times a_{test} \cdot V_{st}}$$

where:

S_{st} – the area of the main peaks in the chromatogram of solutions of RSO of rutin and quercetin, mAU;

S_{test} – peak area of rutin and quercetin on the chromatogram of the test sample, mAU;

a_{st} – weighed RSO of rutin and quercetin, g;

P – the content of rutin and quercetin in the standard sample, %.

Determination of vitamin K₁. Liquid chromatograph with isocratic pump and variable wavelength spectrophotometric detector, type Agilent 1200 series or equivalent. Chromatographic column with a size of 3.0x150 mm, filled with Zorbax Eclipse XDB C-18 sorbent, with a particle size of 3.5 μm or similar; mobile phase: methanol: acetonitrile (55:45%/%) filtered and degassed in any convenient way; detection wavelength - 262 nm; mobile phase flow rate - 1.0 ml/min; temperature - room; the volume of the injected sample - 20 μl; the analysis time should be at least 3 times the retention time of the main peak (vitamin K₁). Alternately injected into the chromatograph injector, 20 μl of solution A, B each, and chromatograms were obtained. The content of related compounds was evaluated by comparing the peak areas in the chromatograms obtained with solutions A and B. The limits of the content of related compounds:

- the area of the vitamin K₁ peak on the chromatogram of solution A should not exceed the sum of the areas of all peaks on the chromatogram of solution B by more than 2 times (no more than 10.0%);
- the peak area of the related compound A in the chromatogram of solution A should not exceed the sum of the areas of all peaks in the chromatogram of solution B (not more than 5.0%);
- the area of the peaks of individual impurities in the chromatogram of solution A should not exceed the sum of the areas of all peaks in the chromatogram of solution B by more than 0.4 times (no more than 2.0%); the sum of the peak areas of unidentified impurities in the chromatogram of solution A should not exceed the sum of the areas of all peaks in the chromatogram of solution B by more than 1.4 times (no more than 7.0%); do not take into account peaks with areas less than 0.04 of the sum of the areas of all peaks in the chromatogram of solution B (0.2%).

Preparation of a solution of RSO rutin. 0.0055 g of rutin (FS 42 Uz-0137-2013) was dissolved in methanol in a pycnometer with a capacity of 10 ml and the volume was adjusted to the mark with the same solvent.

Preparation of a solution of RSO quercetin. 0.0054 g of quercetin (No. UA/0119/0101 23.08.2013) was dissolved in methanol in a pycnometer with a capacity of 10 ml and the volume was adjusted to the mark with the same solvent.

Preparation of the test solution. About 1 g (accurately weighed) of the gel was placed in a 100 ml flask, and 10 ml of 70% ethanol was added. After complete dissolution, it was filtered through a Millipore brand filter.

Preparation of a solution of RSO vitamin K₁. About 0.50 g (accurately weighed) vitamin K₁ was transferred into a 100 ml volumetric flask, dissolved in the mobile phase and the solution volume was brought up to the mark. 2.5 ml of solution A was transferred into a 50 ml volumetric flask and the solution volume was adjusted to the mark. B).

Preparation of the test solution (vitamin K₁). About 2.5 g (accurately weighed) of the drug was transferred into a volumetric flask with a capacity of 100 ml, dissolved in the mobile phase and the volume of the solution was brought up to the mark. a solution of 2.5 ml of solution A was transferred into a volumetric flask with a capacity of 50 ml and the volume of the solution was brought up to the mark fast solution B).

3. Results and Its Discussion

Chromatograms of the test sample are shown in Figures 1, 2. Chromatograms of working standard samples of vitamin K₁, rutin.

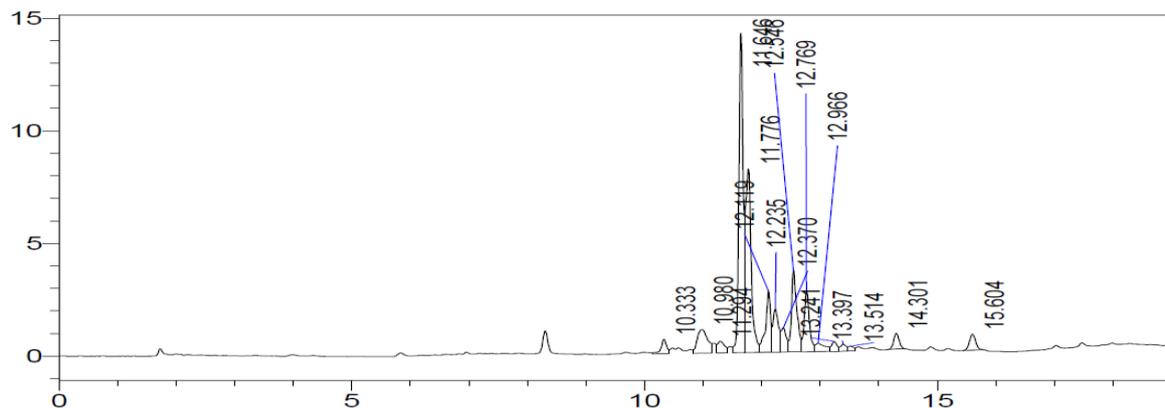


Figure 1. Gel chromatogram (rutin, quercetin)

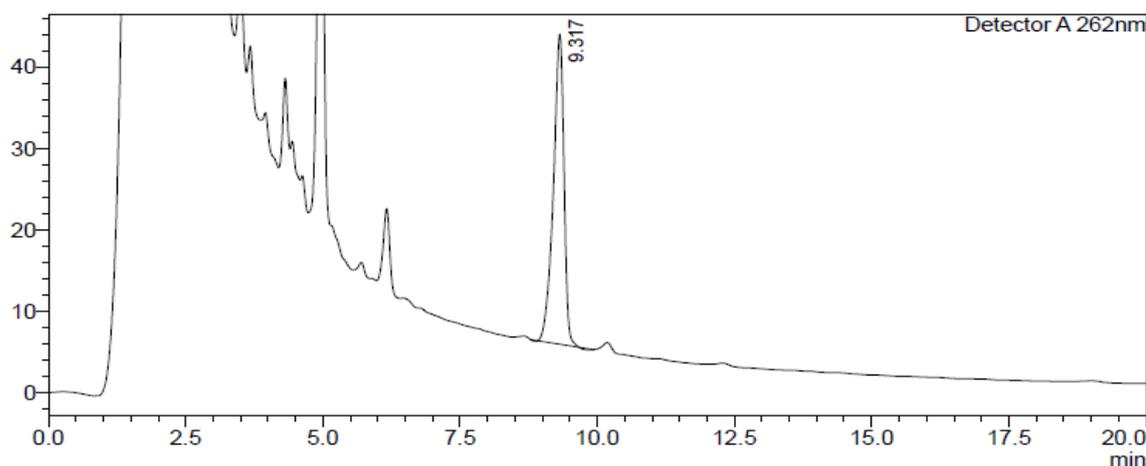


Figure 2. Gel chromatogram. (vitamin K₁)

Table 5. Statistical processing of the results of the analytical gel technique (total flavonoids)

Statistical characteristics	Results
Lowest value	0,050±0,001 mg/ml
Highest value	0,051 mg/ml
Average value	0,0506 mg/ml
Standard deviation	0,0007±0,00001
The coefficient of variation	0,0044±0,0001
Confidence interval (at P=95%)	2,78±0,002
Relative error of definition	3,68±0,011%

Table 6. Statistical processing of the results of the analytical gel technique (vitamin K₁)

Statistical characteristics	Results
Lowest value	0,0167±0,0002 mg/ml
Highest value	0,0168 mg/ml
Average value	0,01678 mg/ml
Standard deviation	0,000447
The coefficient of variation	0,00278
Confidence interval (at P=95%)	2,78±0,002
Relative error of definition	2,3±0,01 %

From figures 1 and 2 it follows that, on the chromatograms of the resulting gel, peaks were retained corresponding to the peaks of standard samples of quercetin (14.3 min.), Rutin (11.7 min.) and vitamin K₁ (9.3 min.). Statistical results for the sum of flavonoids and vitamin K₁ are presented in tables 5 and 6.

4. Conclusions

A dental gel was developed based on local plant materials, and bases for the gel were also selected. As excipients selected: sodium carboxymethylcellulose (NaCMC - the basis of the gel), nipagin as a preservative, purified water and glycerin.

To study biologically active substances in the gel, HPLC methods have been developed for the quantitative determination of flavonoids and vitamin K₁. The content of rutin was 0.051 mg/ml, and vitamin K₁ was 0.0168 mg/ml. The relative error in the content of rutin was 3.68%, vitamin K₁ 2.3%.

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