

Physicochemical Characterization and Thermal Properties of Lipids from *R. opacus* PD630

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Abstract Physicochemical properties of lipid from *R. opacus* PD630 were studied along with positional distribution of triacylglycerol fatty acids and thermal properties of the oil. Results showed that saturated fatty acids (SFA) contribute a large portion ($51.50 \pm 0.49\%$) of total fatty acids. A high proportion of fatty acids with an odd number of carbon atoms ($32.08 \pm 0.39\%$) were also detected. The lipid was high in red and yellow colour parameters and contained a high level of unsaponifiable matter. It was oxidatively stable and had good quality based on its moisture and volatile content, peroxide value, *p*-anisidine value, TOTOX value, conjugated diene and triene values, free fatty acid and acid values. The SFA ($70.56 \pm 1.95\%$) were enriched at *sn*-2 position against *sn*-1,3 position while unsaturated fatty acids were predominantly found at *sn*-1,3 position. The melting and crystallization of lipid from *R. opacus* PD630 occurred over a large temperature range from -7.45°C to 32.37°C and 17.62°C to -19.93°C respectively, due to its high degree of saturation. Lipids from *R. opacus* PD630 decomposed in four stages during programmed heating, comprising decomposition of unsaponifiable matter, monounsaturated fatty acids (MUFA) and SFA degradation, followed by decomposition of MUFA and SFA degradation products. Based on the data, lipid from *R. opacus* PD630 can be considered as a good quality oil and is thermally stable.

Keywords Physicochemical properties, Thermal properties, Oil quality, *R. opacus* PD630

1. Introduction

Microbial oils have been receiving increasing attention as a source of novel oils. Microorganisms accumulating more than 20-25% of their biomass as oil may be termed as oleaginous and their oils as single cell oils, unicellular oils or microbial oils. Microbial oils have a number of potential commercial applications as nutraceuticals, pharmaceuticals, feed ingredients for aquaculture as well as feedstock for producing biodiesel [1]. The major hindrance to their commercialisation is their high cost of production.

Rhodococcus opacus strain PD630 is an oleaginous microorganism species, which has the ability to synthesize a large percentage of oil. In *R. opacus* PD630, the triacylglycerols (TAG) are synthesized in the cytoplasm as insoluble inclusions from several carbon sources such as carbohydrates, alkanes or fatty acids. They can accumulate up to 76% of the cellular dry matter, probably the highest TAG content ever found in bacteria. The lipid from *R. opacus* PD630 was reported to contain an unusual fatty acid composition. The TAG was composed of fatty acids with 14-18 carbon atoms with a large portion of odd numbered

fatty acids [2, 3]. Recent studies have shown the anticarcinogenic effects of odd- and branched-chain fatty acids on cancer cells and antifungal effect of odd numbered fatty acids [4, 5].

Information on thermal properties is important in lipid analysis to provide information about functionality and application of the lipids [6]. Differential Scanning Calorimetry (DSC) is a thermal technique to measure heat flow and heat capacity of lipid samples during phase transitions, including melting and crystallization, which are associated with chemical composition [6]. Thermal decomposition can be monitored by thermogravimetric analysis (TGA) through weight change under heating. A higher temperature of decomposition suggests a higher thermal stability [7].

The biotechnological relevance of TAG is due to the production of novel lipids that are different to those already available from animals and crops [8]. *R. opacus* PD630 can become an interesting candidate for the biotechnological production of "high-value single-cell lipid" as well as other novel lipids such as branched-chain fatty acids that can be utilized for food technology and other specialized applications if genetically modified strains of this species are applied [9].

The aim of this study was to characterize the physicochemical properties of lipid from *R. opacus* PD630, to determine positional distribution of fatty acids in the lipid

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and to study its thermal properties.

2. Materials and Methods

2.1. Materials

R. opacus PD630 (DSM 44193) cells were prepared in the Department of Microbiology and Immunology, University of Otago, New Zealand. Solvents and reagents were all of analytical grade.

2.2. Methods

2.2.1. *R. opacus* PD630 Cultivation and Harvest

R. opacus PD630 cultivation and harvest was conducted according to a modified [10] method. The culture media used were brain heart infusion (BHI) broth and a phosphate-buffered defined medium containing 400 g sucrose, 21 g (NH₄)₂SO₄, 10 g MgSO₄·7H₂O, 113.3 mg CaCl₂, 5 ml antifoam, 0.35 mol phosphate buffer, 10 ml trace element solution, and 10 ml stock A solution per liter. Phosphate buffer, stock a solution and trace element solution were the same as those described [11].

BHI broth was inoculated with *R. opacus* PD630 and incubated at 30°C on a shaking incubator at 200 rpm for 72 hours. Cells were harvested by centrifugation at 4000 rpm for 15 minutes, resuspended in sterile saline, and added into sterilized phosphate-buffered defined medium. The mixture was added into the fermentor (BioFlo 410; New Brunswick Scientific Inc., USA) and cultivated at 30°C with ambient air supply at 10 L/m (1 vvm), pH 7.0 control with 2 M NaOH and shaking (200 rpm) for 72 hours. At the end of the cultivation, biomass was collected by centrifugation at 4000 rpm for 15 minutes. The retained wet cell mass was resuspended in 500 ml of reverse osmosis water.

2.2.2. Lipid Extraction

The crude lipids were extracted by mixing chloroform-methanol (1:1 v/v) with the resuspended *R. opacus* PD630 cells in the proportion 2:1 according to a modified method [12]. The mixtures were centrifuged at 3000 rpm for 10 minutes and the chloroform layer was collected and evaporated by rotary vacuum evaporator to obtain the crude lipid.

2.2.3. Fatty Acid Composition

Fatty acids were analyzed as fatty acid methyl esters (FAME) according to a modified [13] method. Lipid sample (50 mg) and internal standard (C10:0; 1 mg) was mixed with KOH in methanol (0.5N; 2 ml) and saponified at 80°C for 2.5 hours. Diethyl ether (2 ml) and 5 ml of milli-Q water were added into the mixture. The aqueous phase was acidified with hydrochloric acid (37%) until it was red to litmus paper. The diethyl ether phase (upper phase) was collected and added into 2ml of boron trifluoride (14%) in methanol and heated at 80°C for 20 minutes. Saturated NaCl (5ml) solution

was added into the mixture. The upper phase containing FAME was collected for gas chromatography (GC) analysis. A BP70 capillary column (50 m × 330µm × 0.25µm) was used for separation of the FAME using gas chromatography-flame ionization detection (Agilent 6890N; Agilent Technologies Inc., USA) with a split ratio of 20:1. Hydrogen was used as the carrier gas and the flow rate was 2.2 ml/min. The temperatures of injector and detector were 250°C.

The temperature settings were as follows: 70°C to 230°C at 5°C/min and isothermal at 230°C for 20 minutes. The fatty acids were identified using a FAME reference standard (FAMQ-005). HP Chemstation computer software was used for data analysis.

2.2.4. Triacylglycerol Positional Distribution of Fatty Acids

Analysis of positional distribution of fatty acids was conducted according to a modified method [14]. Lipid sample (100mg) was mixed with 3 ml of buffer containing 1M of tris (hydroxymethyl)-amino methane pH8.0, 0.15M CaCl₂, and 0.01% bile salts in a test tube. Pancreatic lipase, Type II (200 mg) was mixed with 5ml tris buffer, and 0.5ml of this mixture was added into the test tube. The test tube was incubated at 37°C on a shaking incubator at 250 rpm for eight minutes. Diethyl ether (3 ml) was added for phase separation. The diethyl ether phase was transferred into a new test tube, back-washed with 2 ml of milli-Q water and concentrated under a stream of nitrogen gas.

The concentrated diethyl ether and the neutral lipid standards were spotted on a glass silica gel thin layer chromatography (TLC) plate (silica gel 60 F₂₅₄) which was previously washed with hexane-diethyl ether (50:50, v/v). Separation of lipid classes was conducted using a hexane-diethyl ether-acetic acid (50: 50: 1, v/v/v) solution. The plate was left to dry at room temperature when the solvent had moved up to 90% to the top of the plate. The TLC plate was sprayed with 5% phosphomolybdic acid in ethanol and placed under UV light to observe the lipid fractions. The monoacylglycerol and free fatty acid bands were marked and scraped off the plate for the methylation process and analysis by GC as outlined in section 2.2.3.

2.2.5. Colour

Analysis of oil colour was carried out using a MiniScan XE spectrocolourimeter (HunterLab) with a measuring head hole of 22mm, D₆₅ colour illuminant and 10° observer. Colour intensities were measured using the CIELAB (L*, a*, b*) colour scale.

2.2.6. Determination of Physical and Chemical Properties

Moisture and volatile matter content, unsaponifiable matter content, peroxide value, *p*-anisidine value, free fatty acid content, and acid value were determined at 25°C according to [15] methods. Specific extinctions of lipids were determined spectrophotometrically at 230 and 270 nm as outlined [16].

2.2.7. Thermal Analysis

Melting and crystallization characteristics of lipid from *R. opacus* PD630 were measured by DSC (Q2000, TA instruments Ltd., USA) using sealed aluminum pans (Tzero pan and lid) with a sample of 10mg. The experiments were carried out over the temperature range -50°C to 60°C at 2°C/min under nitrogen atmosphere (50 ml/min). The DSC instrument was calibrated using indium (melting point = 156.6°C, enthalpy = 28.45 J/g) according to the DSC instrument manual. Melting and crystallization profiles, transition enthalpies and solid fat content (SFC) of the lipid sample were produced using TA Universal Analysis 2000 software.

The thermal decomposition process was monitored by TGA (Q50, TA instruments Ltd., USA). Samples (10 to 15 mg) were heated from 10 °C to 700°C at 2°C/min in an air atmosphere. Data on the percentage of weight change were obtained using the TA Universal Analysis 2000 software. The onset temperature for decomposition was extrapolated from the thermal decomposition curve according to [7].

2.2.8. Statistical Analysis

All measurements were carried out in triplicate. Data were reported as mean \pm standard deviation (SD). Statistical analysis of data was performed using IBM SPSS Statistics Version 20. One way ANOVA and Tukey's test were used to determine significant difference between samples for each measurement at $p < 0.05$.

3. Results and Discussion

3.1. Fatty Acid Composition

Table 1. Fatty acid composition of lipids extracted from *R. opacus* PD630

Fatty acids	Concentration (mg/g)	Percentage (mol %)
14:0	12.25 \pm 0.21 ^h	1.77 \pm 0.03 ^h
15:0	35.42 \pm 1.18 ^g	4.81 \pm 0.16 ^g
16:0	218.8 \pm 1.17 ^a	28.10 \pm 0.15 ^a
17:0	94.70 \pm 0.66 ^d	11.53 \pm 0.08 ^d
18:0	45.68 \pm 0.60 ^f	5.29 \pm 0.07 ^f
ΣSFA	406.9 \pm 3.82	51.50 \pm 0.49
16:1	64.24 \pm 0.31 ^e	8.31 \pm 0.04 ^e
17:1	128.3 \pm 1.39 ^c	15.74 \pm 0.17 ^c
18:1	209.7 \pm 1.63 ^b	24.45 \pm 0.19 ^b
ΣMUFA	402.2 \pm 3.33	48.50 \pm 0.40

Means \pm SD, n = 3.

^{abcdeh}Values with different superscript letters within a column are significantly different ($p < 0.05$)

The fatty acid composition of lipid from *R. opacus*PD630 is shown in Table 1. The most dominant fatty acid was palmitic acid (C16:0), which contributed $28.10 \pm 0.15\%$ of total fatty acids followed by oleic acid (C18:1). The result confirms the relatively large proportion of margaroleic acid (C17:1; $15.74 \pm 0.17\%$) and palmitoleic acid (C16:1; $8.31 \pm$

0.04%) which may give health benefits. SFA contributed $51.50 \pm 0.49\%$ of total fatty acids, while monounsaturated fatty acid (MUFA) contributed $48.50 \pm 0.40\%$. There were no polyunsaturated fatty acids found in lipid from *R. opacus*PD630. A large portion ($32.08 \pm 0.41\%$) of odd carbon numbered fatty acids was found in the lipid. The data agrees with the results from previous studies conducted [2, 3, 17].

3.2. Positional Distribution

The results of positional distribution of the fatty acids are shown in Table 2. The results show the saturated fatty acids (SFA) prefer *sn*-2 position against *sn*-1,3 position while unsaturated fatty acids were predominantly found at *sn*-1,3 position. The results of the present study were similar to previous results [3]. However, there were more MUFA found at *sn*-2 position, but less SFA found at *sn*-1,3 position in the present study due to the different carbon source used for *R. opacus* PD630 cultivation. The positional distribution of the fatty acids in lipids of *R. opacus*PD630 are similar to animal fats as SFA are predominantly found at *sn*-2 position in most animal fats, whereas MUFA prefer *sn*-2 position against *sn*-1,3 position in vegetable oils [18].

Table 2. Positional distribution (%) of fatty acids in *R. opacus* PD630 lipid following pancreatic lipase treatment

Fatty acids	<i>sn</i> -2	<i>sn</i> -1,3
14:0	2.74 \pm 0.14 ^a	0.28 \pm 0.02 ^b
15:0	8.23 \pm 0.21 ^a	0.58 \pm 0.05 ^b
16:0	39.35 \pm 0.99 ^a	8.14 \pm 0.23 ^b
17:0	14.12 \pm 0.43 ^a	1.24 \pm 0.10 ^b
18:0	6.12 \pm 0.22 ^a	1.16 \pm 0.12 ^b
ΣSFA	70.56 \pm 1.95^a	11.40 \pm 0.52^b
16:1	7.17 \pm 0.26 ^b	22.51 \pm 0.60 ^a
17:1	10.00 \pm 0.37 ^b	21.68 \pm 0.33 ^a
18:1	12.27 \pm 0.20 ^b	44.42 \pm 0.83 ^a
ΣMUFA	29.44 \pm 0.83^b	88.61 \pm 1.76^a

Means \pm SD, n = 3.

^{ab}Values with different superscript letters within a row are significantly different ($p < 0.05$)

3.3. Colour

Table 3. CIELAB $L^*a^*b^*$ value[#] of lipid from *R. opacus* PD630

Colour Parameters	<i>R. opacus</i> PD630
L^*	49.33 \pm 0.67
a^*	15.37 \pm 0.48
b^*	19.14 \pm 0.45

Means \pm SD, n=3

[#] L^* represents lightness of the sample, 0 = black, 100 = white; a^* represents redness when positive, greenness when negative; b^* represents yellowness when positive, blueness when negative.

Table 3 shows the CIELAB $L^*a^*b^*$ colour values of lipid from *R. opacus*PD630. The a^* and b^* values of lipid from *R. opacus*PD630 (15.37, 19.14) were higher than that of vegetable oils including olive oil (68.3, 3.1), sunflower oil (65.8, 3.4), corn oil (66.4, 4.1) and palm oil (64.5, 4.0), which suggest that the lipid contained a higher level of

yellow pigments [19]. The L^* value of lipid from *R. opacus* PD630 (49.33) was lower than that of olive oil (9.9), sunflower oil (10.0), corn oil (10.0) and palm oil (9.6), which suggests that lipid from *R. opacus* PD630 was darker compared with vegetable oils [19].

3.4. Quality Characteristics of Lipid from *R. opacus* PD630

Table 4 shows the physicochemical characteristics of lipid from *R. opacus* PD630. The percentage of moisture and volatile matter ($1.08 \pm 0.08\%$) was higher compared to that of vegetable oils (0.60-0.72%) including hemp seed oil, canola oil and flaxseed oil [20]. A large amount ($15.04 \pm 0.66\%$) of unsaponifiable matter content was found in the oil sample. It is desirable for good quality oil to have low moisture and unsaponifiable matter content [20].

Table 4. Physicochemical characteristics of oil from *R. opacus* PD630

Parameters	<i>R.opacus</i> PD630
Moisture and volatile content (%)	1.08 ± 0.08
Unsaponifiable matter (%)	15.04 ± 0.66
Peroxide value	0.07 ± 0.00
(mequiv. peroxide/kg sample)	
<i>p</i> -Anisidine value (unit)	0.31 ± 0.04
TOTOX value (unit)	0.45 ± 0.04
Conjugated diene (unit)	0.07 ± 0.00
Conjugated triene (unit)	0.09 ± 0.00
Free fatty acids, as oleic (%)	1.25 ± 0.03
Acid value (mg KOH/g of sample)	2.48 ± 0.04

Means \pm SD, n=3

Peroxide value measures the primary oxidation products in lipid. The result shows that the peroxide value of lipid from *R. opacus*PD630 was 0.07 ± 0.00 mequiv. peroxide/kg sample, which indicates the oil sample maintained good quality as it did not exceed the limit of 10 under the New Zealand Food Regulations [21]. *p*-Anisidine value measures the secondary oxidation products that resulted from decomposition of hydroperoxides. The *p*-anisidine value (0.31 ± 0.04) of lipid from *R. opacus* PD630 was within the limit of 2 indicating good quality [22]. The TOTOX value is a measurement of overall oxidation. This value indicated that lipid of *R. opacus*PD630 was good in quality as the TOTOX value (0.45 ± 0.04) was less than 4 [23].

Conjugated diene and triene values represent the concentration of conjugated hydroperoxides in oils resulting from shifting of double bonds [24]. The conjugated diene value (0.07 ± 0.00) of lipid from *R. opacus* PD630 was much lower than that of most vegetable oils including peanut oil (0.37), hemp oil (1.53), flax oil (2.08) and canola oil (2.21) whereas the conjugated triene value (0.09 ± 0.00) of lipid from *R. opacus*PD630 was slightly higher compared to these vegetable oils (0.02-0.07) [20, 25].

The percentage of free fatty acids and acid value in lipid of *R. opacus*PD630 might be related to the moisture content in lipid, as the moisture in lipid leads to hydrolytic breakdown of TAG to form free fatty acids. The free fatty acid value of lipid from *R. opacus*PD630 was $1.25 \pm 0.03\%$ and the acid

value was 2.48 ± 0.04 mg KOH/g. The quality of the lipid from *R. opacus* PD630 was good as the acid value was lower than 4.0mg KOH/g [21].

3.5. Thermal Analysis

Figure 1 shows the melting profile of lipid from *R. opacus* PD630. There were two endothermic regions of the melting curve. The first endothermic region at $1.69 \pm 0.33^\circ\text{C}$ (Table 5) represents the melting of MUFA and the second endothermic region at $26.59 \pm 0.46^\circ\text{C}$ indicates the contribution of SFA. The peak shapes are the result of overlapping effects from composition and polymorphism [26]. The total melting enthalpy for lipid from *R. opacus* PD630 was $47.50 \pm 2.96\text{J/g}$. The melting enthalpy of the second region ($37.72 \pm 3.72\text{J/g}$) was higher than that of the first region ($9.35 \pm 1.20\text{J/g}$). This is due to unsaturated fatty acids not aligning as neatly as SFA due to their *cis* double bonds, hence requiring less energy to overcome the intermolecular attraction, and melting first.

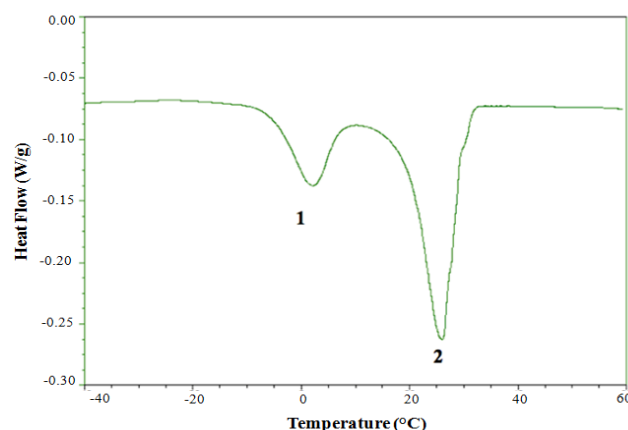


Figure 1. DSC melting profile of lipid from *R. opacus* PD630

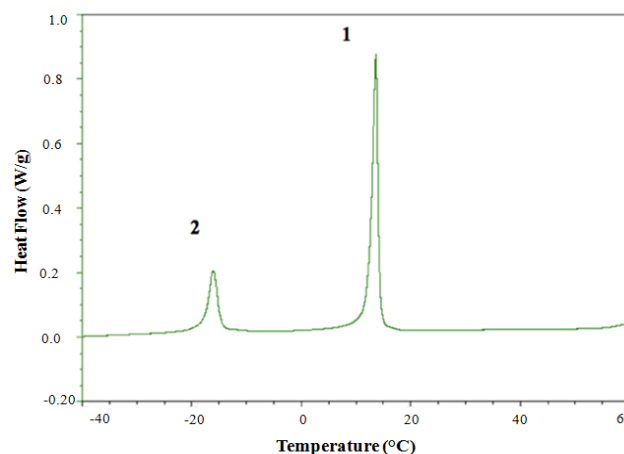


Figure 2. DSC crystallization profile of lipid from *R. opacus* PD630

Figure 2 shows the crystallization profile of lipid from *R. opacus* PD630. The crystallization profile of lipid from *R. opacus* PD630 exhibited two exothermic regions. The first exothermic peak occurred at $14.52 \pm 0.67^\circ\text{C}$ (Table 5) and the second exothermic peak took place at $-16.31 \pm 0.34^\circ\text{C}$. The crystallization enthalpy of lipid from *R. opacus* PD630 was $-52.59 \pm 1.99\text{J/g}$. There was no significant

difference in the magnitude of heat involved in melting and crystallization indicating that the melting and crystallization processes corresponded with each other.

The result showed that lipid from *R. opacus* PD630 have a higher melting crystallization range compared to several oils that have been investigated previously, due to a higher degree of saturation [6, 20].

Table 5. Transition temperature and enthalpy during melting and crystallization of lipid from *R. opacus* PD630

Parameters		Melting	Crystallization
Region 1	$T_{\max 1} (^{\circ}\text{C})$	1.69 ± 0.33	14.52 ± 0.67
	$\Delta H_1 (\text{J/g})$	9.35 ± 1.20	-41.28 ± 1.19
Region 2	$T_{\max 2} (^{\circ}\text{C})$	26.59 ± 0.46	-16.31 ± 0.34
	$\Delta H_2 (\text{J/g})$	37.72 ± 3.72	-11.31 ± 0.83
Total $\Delta H_T (\text{J/g})$		47.50 ± 2.96	-52.59 ± 1.99
Temperature range ($^{\circ}\text{C}$)		-7.45 to 32.37	17.62 to -19.93

Mean \pm SD, n=3

$T_{\max 1}$ = Peak maxima of region 1, $T_{\max 2}$ = Peak maxima of region 2, ΔH_T = Melting enthalpy region 1, ΔH_2 = Melting enthalpy region 2, ΔH_T = Total melting enthalpy

Figure 3 illustrates the SFC of lipid from *R. opacus* PD630. The SFC was calculated on the basis of the area under the integration line over the melting curve. The melting of lipid from *R. opacus* PD630 showed two stages. As shown in Figure 3, the solid fat content started to drop rapidly at the beginning, then slowed down at around 5 $^{\circ}\text{C}$, and begin to decrease rapidly again from around 10 $^{\circ}\text{C}$ until there was no solid fat left. The first stage represents the melting of MUFA-rich TAG in the lipid, and the second stage represents the melting of saturated TAG.

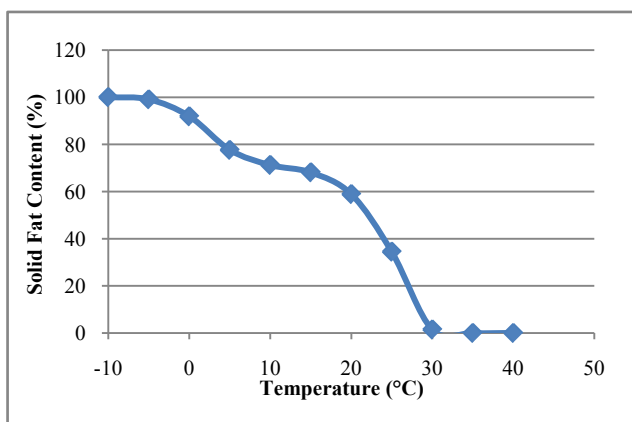
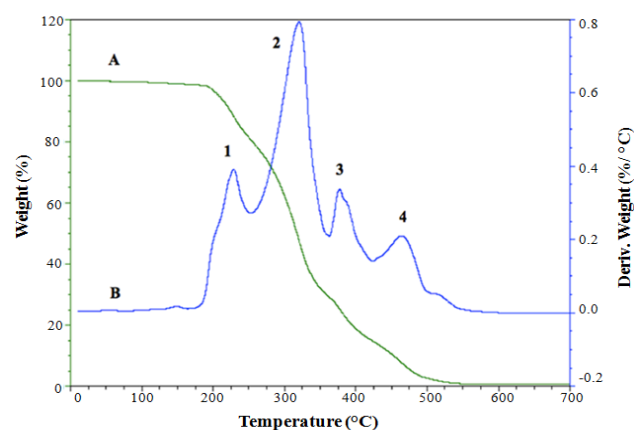


Figure 3. Solid fat content of lipid from *R. opacus* PD630

The thermal decomposition process for lipid of *R. opacus* PD630 from 10 $^{\circ}\text{C}$ to 700 $^{\circ}\text{C}$ by TGA is shown in Figure 4. Curve A and B show the percentage of weight remaining and rate of weight loss of lipid respectively. The result showed that the lipid of *R. opacus* PD630 started to decompose from 185.4 $^{\circ}\text{C}$, and there was no residual remaining after 550 $^{\circ}\text{C}$.

The rate of weight loss of lipid from *R. opacus* PD630 exhibited four stages of lipid decomposition during heating. The first stage (173.6 to 252.1 $^{\circ}\text{C}$) and second stage (252.1 to 362.2 $^{\circ}\text{C}$) represent the decomposition of unsaponifiable

matter content followed by MUFA and SFA respectively. The third stage (362.2 to 423.5 $^{\circ}\text{C}$) and last stage (423.5 to 545.4 $^{\circ}\text{C}$) represent the decomposition of degradation products formed from previous steps. This finding corresponds with previous study on vegetable oils, in which unsaturated fatty acids start to decompose prior to SFA [7].



A: percentage of weight remaining curve

B: rate of weight loss curve

Figure 4. Thermal decomposition of lipid from *R. opacus* PD630 by TGA from 10 $^{\circ}\text{C}$ to 700 $^{\circ}\text{C}$

4. Conclusions

In conclusion, lipid of *R. opacus* PD630 in the present study was less susceptible to lipid oxidation than vegetable oils in previous studies due to the high SFA content, as reflected by its low peroxide value, *p*-anisidine, TOTOX value, free fatty acid content, acid value, conjugated diene and triene value. However the unsaponifiable matter content of lipid is much higher than most animal and vegetable lipids. The colour parameters suggest that the lipid from *R. opacus* PD630 was darker, redder and more yellow compared with vegetable oils. In TAG of *R. opacus* PD630, SFA were predominantly found at *sn*-2 position and unsaturated fatty acids tend to prefer the *sn*-1,3 position against the *sn*-2 position, which is similar to animal fat. The lipid of *R. opacus* PD630 showed a wide melting and crystallization range. Thermal decomposition of lipid from *R. opacus* PD630 exhibited four stages that represent the decomposition of unsaponifiable matter content, MUFA, SFA and their degradation products. Based on the above data, lipid from *R. opacus* PD630 can be considered as good quality oil which is also thermally stable. The unusually high content of margaric and margaroleic acid warrants further investigation for potential health benefits.

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