

Synthesis, Reactions and Biological Importance of α , β -Unsaturated Carbodithioate Esters: A Review

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Abstract Carbodithioate esters are important functional organosulfur compounds widely used in different fields such as pharmaceuticals, agrochemicals and material sciences. α , β -unsaturated dithioesters, carbodithioates are the type of organosulfur chemical compounds that attract the special interest of organic chemists. α , β -unsaturated dithioester type compounds have various applications in medicinal chemistry and provided numerous potent α , β -unsaturated dithioester derivatives for different therapeutic targets. α , β -unsaturated dithioesters are widely used as solvents, polymers, and biopharmaceutical agents. For the treatment of tuberculosis, leprosy and dermatitis herpetiformis diseases, several drug molecules containing α , β -unsaturated dithioester groups are used. Based on these applications, researchers have engaged in the preparation and study of many types of α , β -unsaturated dithioester derivatives for their medicinal activities e.g. biological, antimalarial, antimicrobial, anti-inflammatory, anticancer, anti-HIV, and anti-inflammatory properties. The present article provides a targeted review of recent synthetic strategies, pertinent reactions and applications of α , β -unsaturated dithioester type compounds to facilitate future research efforts so that the medicinal and industrial applications of this important class of compounds continue to be beneficial to society.

Keywords α , β -unsaturated carbodithioates, Synthesis, Anticancer activity, Biological importance, Medicinal chemistry

1. Introduction

Carbon-sulfur bond formation is a fundamental approach to introduce sulfur into organic compounds. Carbon-sulfur bond formation has received considerable attention due to the prominence of the C-S bond in various molecules that are of biological, pharmaceutical and material interest [1].

α , β -unsaturated thioesters have attracted much attention as active esters for the syntheses of different compounds. Synthetic methods for α , β -unsaturated dithioesters have received considerable interest in view of their increased reactivity, compared to their carboxylic analogues, as potential dienes or dienophiles in hetero Diels–Alder cycloadditions [2]. Moreover, the cycloaddition products, thiochromenes, are potential precursors of a wide range of thioheterocycles with interesting biological properties. There are very few general methods available for the synthesis of α , β -unsaturated dithioesters and those known are mostly specific to certain substrate classes. The methods available in

the literature include (i) alkylation of thiolate anions obtained by the addition of vinyl cuprates to carbon disulfide [3], (ii) isomerization of α , β -unsaturated dithioesters [4], (iii) base catalyzed elimination of β -hydroxy dithioesters [5], and (iv) Wittig–Horner, Peterson or Mukaiyama type condensation reactions of aldehydes and ketones [6]. Hartke *et al.* have also shown that α , β -unsaturated amides can be transformed into the corresponding dithioesters by a sequence of reactions involving thionation, alkylation and sulfhydrolysis [7]. Although dithioesters have been known for many years [8], it is only recently that α , β -unsaturated dithioesters have attracted attention. We were interested in that α , β -unsaturated dithioesters as potential heterodienes or in cycloaddition reactions [9].

2. Synthesis and Reactions

In 1978, the preparation of the β -hydroxydithioester **1** was described [10] involving the treatment of ethyl dithioacetate with lithium di-isopropylamide (LDA) in tetrahydrofuran (THF) followed by isobutyraldehyde at - 78°C. It is recently that α , β -unsaturated dithioesters have attracted attention. Preparative approaches to these compounds **2** which have been investigated include: (a) reaction of a vinyl cuprate with carbon disulphide followed by methyl iodide [11]; this

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is successful for compounds (**2a-c**), but attempts to make the dithioester (**2d**) in this way gave its dimer; (**b**) sulphhydrolysis at -75°C of the immonium salt derived by S-methylation of the thioamide gave the phenyl derivative (**2e**) [12] which dimerized above -30°C ; (c) base-catalysed isomerisation of β,γ -unsaturated dithioesters [13], in turn prepared from N-phenyliminothioesters, gave (**2a**) and, at -40°C , (**2f**) which dimerized at room temperature; and flash pyrolysis of the bridged anthracene and trapping of the product in a matrix at -196°C gave the parent dithioacrylate (**2g**). K. R. Lawson *et al.* were interested in α , β -unsaturated dithioesters as potential heterodienes or heterodienophiles in cycloaddition reactions [14], and they describe their preparation from β -hydroxydithioesters and some cycloadditions in which they are involved. Subsequent to the completion of their work it was reported that addition of Grignard reagents to β -ketodithioesters gave 3,3-disubstituted β -hydroxydithioesters, and the latter were dehydrated to 3,3-disubstituted α , β -unsaturated dithioesters on treatment with toluene-*p*-sulphonic acid in benzene at reflux [15].

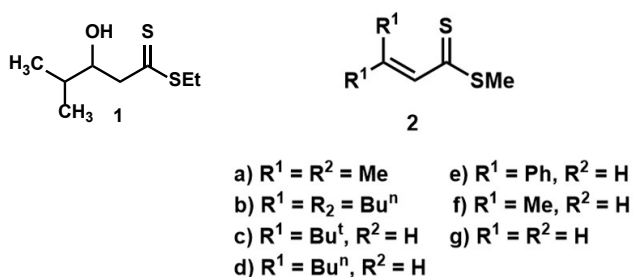
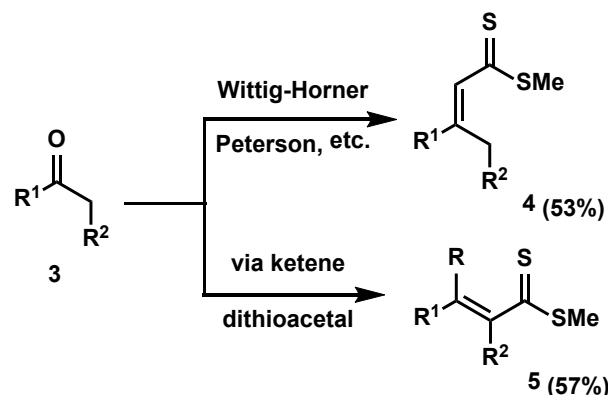


Figure 1. β -hydroxydithioester and α , β -unsaturated dithioesters

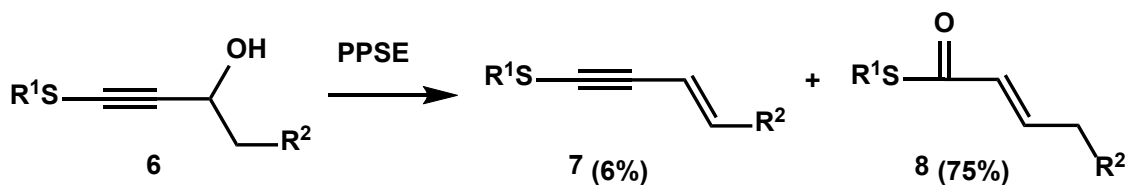
In most of the known methods [6] of preparation of an α , β -unsaturated dithioester starting from carbonyl compound **3**, the dithioester functionality is introduced along with the α -methylene group to form **4**. The present method provides an opportunity to introduce dithioester functionality at the α -position of the carbonyl group of the starting ketone to afford **5** (Scheme 1).



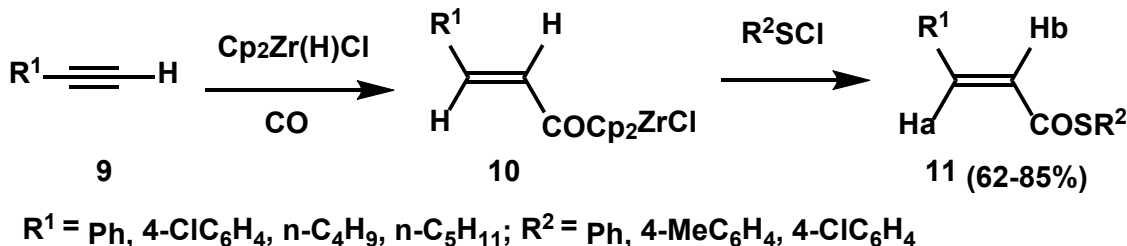
Scheme 1. Synthesis of α , β -unsaturated dithioesters through Wittig-Horner method and via ketene dithioacetal process

There are many ways to synthesize α , β -unsaturated thioesters and dithioesters. The treatment of γ -chalcogen-substituted propargyl alcohols with polyphosphoric acid trimethylsilyl ester (PPSE) gave α , β -unsaturated thioesters via the Meyer-Schuster type rearrangement [16] instead of γ -chalcogen-substituted enynes (Scheme 2). γ -Sulfur-substituted propargyl alcohols **6** reacted with PPSE **7** to give the α , β -unsaturated thioesters **8** in good yields. However, the reactions also gave the enyne sulfides [17].

Coupling reactions of acylzirconocene chlorides with organic halides afforded the corresponding ketones [18]. Considering the high electrophilicity of arylsulfenyl chlorides, P. Zhong and his coworkers [19] attempted to react them with the α , β -unsaturated acylzirconocene chlorides **10**. Experimental results show that, $\text{Cp}_2\text{Zr}(\text{H})\text{Cl}$ [20] adds to terminal alkynes **9** in CH_2Cl_2 at room temperature stereospecifically with high regioselectivity to yield vinylic Zr^{IV} complex, which was stirred under CO atmosphere to give the α , β -unsaturated acylzirconocene chlorides the adducts **10**. **10** react with arylsulfenyl chlorides [21] rapidly at 0°C to afford α , β -unsaturated thioesters **11** with good to excellent yields (Scheme 3).



Scheme 2. Meyer-Schuster rearrangement of γ -sulfur-substituted propargyl alcohols to synthesis of α , β -unsaturated thioesters in 1995

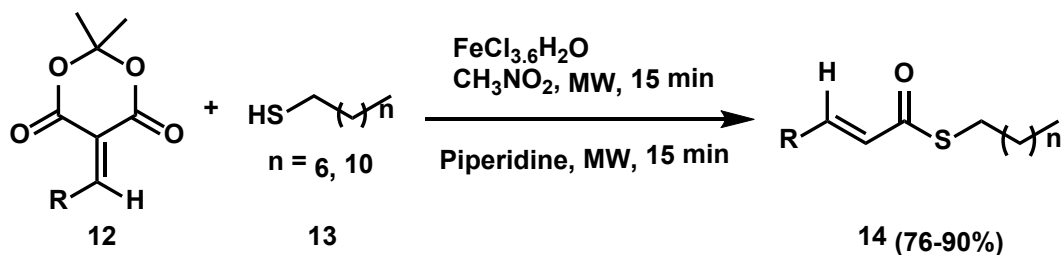


Scheme 3. A stereoselective synthetic route to (E) - α , β -unsaturated thioesters

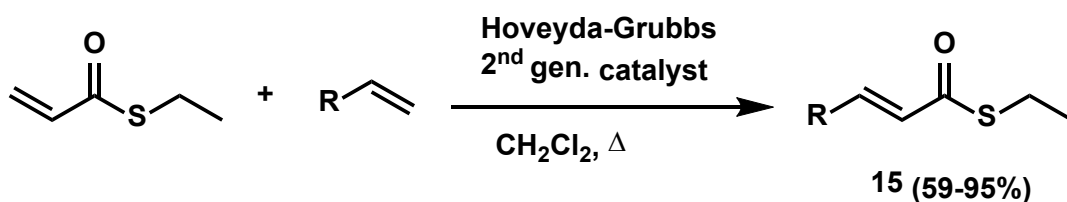
Interestingly, α , β -unsaturated thioesters have marked reactivity as Michael acceptors and they are proved to be excellent substrates in the synthesis of several natural products [22]. Although, it is a very useful intermediate, traditional syntheses of thioesters are encountered with the occasional difficulties such as 1,4-addition of thiolate and subsequent separation from the main product [23]. Olefin cross-metathesis has been elegantly explored to construct α , β -unsaturated thioesters using thioacrylate [24]. Encouraged by the success of synthesis of α , β -unsaturated esters, A.R. Mohite *et al* [25], planned to extend the protocol for the straightforward synthesis of α , β -unsaturated thioesters using the optimized reaction conditions for esters. To compare the reactivity and to extend the application, thiols **13** (1 eq.) were treated with few benzylidene derivative of MA **12** under optimized reaction conditions (Scheme 4). The corresponding α , β -unsaturated thioesters **14** were obtained in good to excellent yields (76–90%) in just 30 min (Scheme 4).

Thioesters are highly relevant compounds due to their distinctive chemical properties: the reduced electron delocalization provides for enhanced reactivity compared to

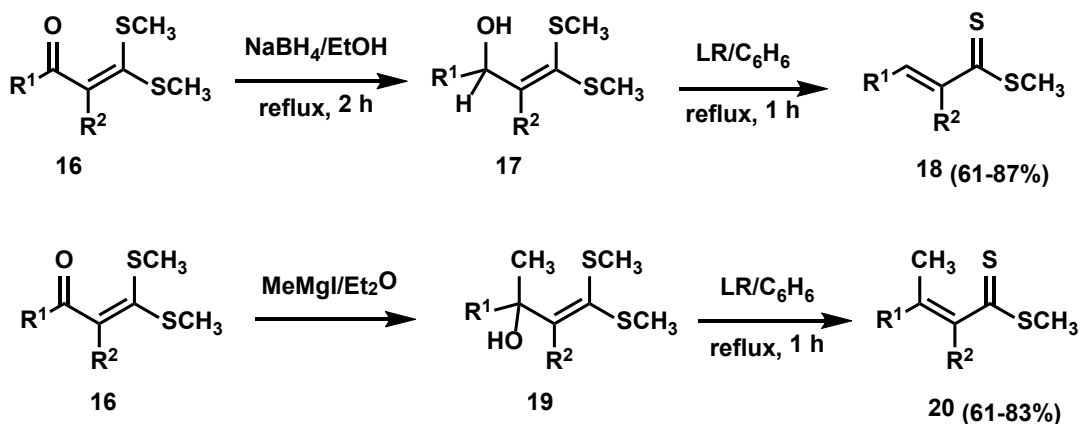
oxoesters [26]. The importance of thioesters in the cell is well established: biological systems use their relative reactivity in many enzymatic reactions by employing, for example, acetyl coenzyme A, cysteine proteases, or polyketide and fatty acid synthases [27]. Their enhanced reactivity compared to that of oxoesters has been employed successfully in a wide range of synthetic organic transformations, some inspired directly by related biosynthetic pathways. Stereoselective aldol reactions often depend on the distinctive reactivity of thioesters [28] and their synthetic versatility is further illustrated by many other well-known transformations including α -alkylations, [29] selective reductions [29, 30], and Pd-catalyzed coupling reactions [31] among others [32]. Considering these importance, A. W. van Zijl and his coworkers [33] found a mild and scalable new route to *S*-ethyl thioacrylate **15** (Scheme 5). The feasibility of the use of this olefin in cross-metathesis reactions with the Hoveyda-Grubbs second generation catalyst is demonstrated. The high functional group tolerance of the reaction allows the preparation of a broad range of versatile functionalized α , β -unsaturated thioesters.



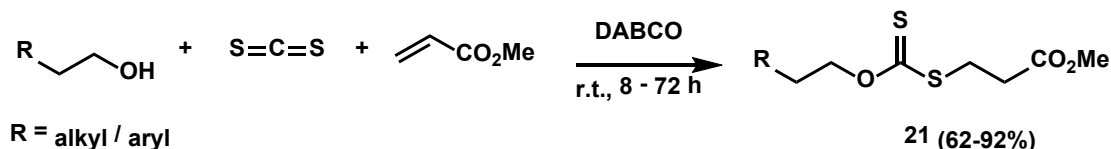
Scheme 4. One-pot direct synthesis of α , β -unsaturated thioesters



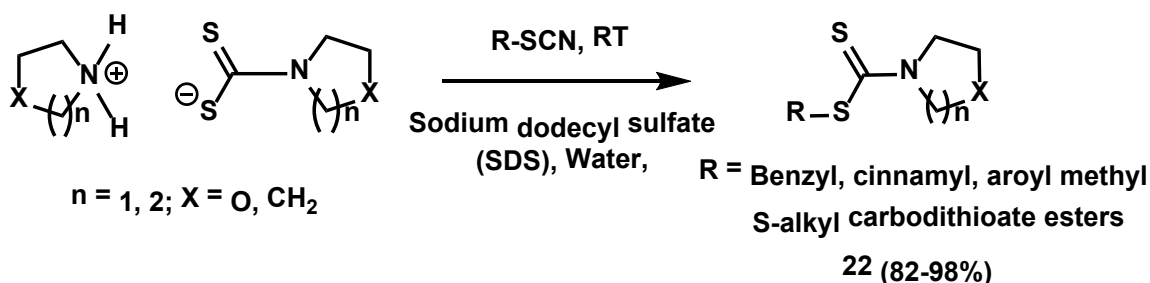
Scheme 5. Cross-metathesis reaction of *S*-ethyl thioacrylate with a variety of olefins to give substituted α , β -unsaturated thioesters



Scheme 6. Synthesis of α , β -unsaturated dithioesters from the reactions of α -hydroxyketene dithioacetals with Lawesson's reagent



Scheme 7. Synthesis of functionalized O, S-dialkylthiocarbonates



Scheme 8. Synthesis of carbodithioate esters from organyl thiocyanates

S. K. Nair and his coworkers have developed a facile two-step process for the conversion of a α -oxoketene dithioacetals to α , β -unsaturated dithioesters, which are valuable intermediates in organic synthesis and the method described here provides a valuable alternative to the previous methods for the synthesis of these compounds. The α -hydroxyketene dithioacetals **17** and **19**, obtained from α -oxoketene dithioacetals **16** by 1,2-reduction or 1,2-addition of carbon nucleophiles, on treatment with Lawesson's reagent afforded α , β -unsaturated dithioesters **18** and **20** in good yields (Scheme 6) [34].

O, S-dialkylthiocarbonates are a class of organo-sulfur compounds which are frequently used as versatile source of radicals [35] and useful intermediates in the synthesis of thiols [36], thiocarbonates [37] alkenes [38], alkanes [39], α , β -unsaturated esters through S-activated carbanions [40] and as photosensitizer [41] of vinyl monomers. Besides, these are used as vulcanization accelerators [42] and in the syntheses of ionic liquids [43]. These are also used to prepare S-containing natural products [44] and find use in Claisen rearrangements leading to interesting derivatives [45].

Normally these (*viz.* dithiocarbonates) are prepared from a three-step process from alcohol, alkyl halide and CS_2 using a strong base [46]. Recently efficient one-pot processes of their preparation have been reported using basic resin (Amberlite IRA) [47] or Trion B [48]. But in those communications only a non-functional alkyl group was used for alkylation of the sulfur center, and the syntheses are stepwise processes.

Multicomponent reactions (MCRs) [49] involve combination of three or more starting materials in a single operation and are gaining popularity in the synthesis of complex compounds due to their high atom economy [50], synthetic convergence and reduced effort in preparation and workup [51]. The early MCRs were mostly discovered by chance or serendipity. But rational design strategies for these reactions are currently being devised [52]. G. C. Patra *et al.* [53] has developed an easy and effective preparation of dithiocarbonates **21** in which the S-alkyl part is

functionalized with an ester or nitrile group employing a three-component single step procedure (Scheme 7).

An efficient and practical method for the preparation of carbodithioate esters **22** from organyl thiocyanates reported by Biswas, K. *et al.* through the reaction with cyclic amine-based dithiocarbamic acid salts in water [54] (Scheme 8). This type of protocol is found to be applicable in general to various thiocyanates such as benzyl or aroyl methyl or cinnamyl and so on. Some other notable features that there are no by-products such as disulfides, metal- and alkali-free, aqueous conditions, and finally easy and near-quantitative formation of cyclic amine-based dithiocarbamic acid salt which acts as a stable alternative reagent.

3. Biological Importance of α , β -Unsaturated Carbodithioate Esters

There is a malignant disease named acute myelogenous leukemia (AML) which is characterized by an aberrant accumulation of immature myeloid haematopoietic cells [55]. This AML is the most common form of acute leukemia in adults and constitutes approximately 80% of cases [56]. Though the treatments of AML significantly improved the rate of remission, still more than 50% relapse with to a resistant form of the disease. So still there is challenge for this AML chemotherapy [57]. Another group of leukemic cells, (Leukemia stem cells (LSCs)) have shown self-renewal ability as well as the capability to produce heterogeneous leukemia cell populations [58, 59]. It has been considered to play significant role in the initiation and relapse of acute leukemia [60]. LSCs are also considered to be an effective strategy for the treatment and possible cure of AML [61-640]. Also, LSCs are refractory to clinical used chemotherapy drugs, such as nucleoside analogues like cytosine arabinoside and anthracyclines like idarubicin and daunorubicin [65, 66]. So, the effective agents that can selectively eradicate LSCs are urgently needed for the development of new therapies for treatment of leukemia.

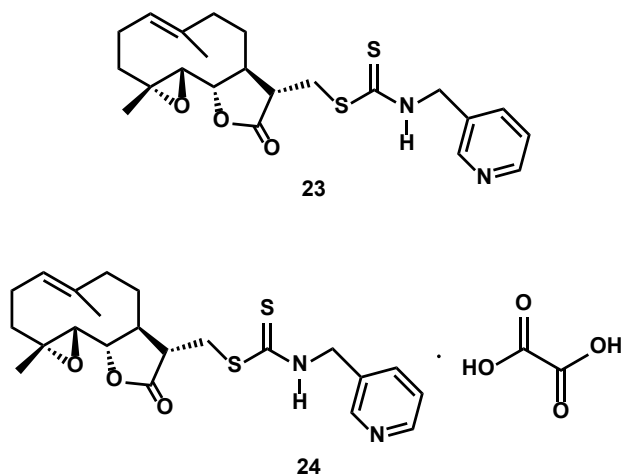


Figure 2. Dithiocarbamate esters **23**, showed potency against AML progenitor cell line KG1a and **24**, a promising drug candidate for the discovery of anti-LSCs drug

Using this information, Dinga, Y. *et al.* designed, synthesized, and evaluated a series of dithiocarbamate esters of parthenolide (PTL) for their anti-AML activities [67]. Among the most promising compound **23** showed greatly improved potency against AML progenitor cell line KG1a with IC_{50} value of 0.7 μ M, and the efficacy found 8.7-folds comparing to that of PTL (IC_{50} = 6.1 μ M). The compound **23** induced the apoptosis of total primary human AML cells and leukaemia stem cell (LSCs) of primary AML cells while sparing the normal cells. The compound **23** suppressed the colony formation of primary human leukaemia cells. The preliminary molecular mechanism study revealed that the compound **23**-mediated apoptosis is associated with

mitogen-activated protein kinase signal pathway. After some of the research results, Dinga, Y. *et al.* proposed that the compound **24** also might be a promising drug candidate for ultimate discovery of anti-LSCs drug (Figure 2).

Dithiocarbamate (*S*-alkyl carbodithioate esters) are the functional organosulfur compounds which were first used as fungicides during World War II [68]. These types of compounds are also largely applied as important fungicides of crops, vegetables and plants [69-71]. Previous reports show that the *S*-alkyl carbodithioate esters and their derivatives show antibacterial [72-74], anticandidal activity and cytotoxicity [75], antihistaminic [76], anticancer properties [77, 78-80] and anthelmintic [77] properties. These types of compounds are very useful in the treatment of cardiovascular disorders and inflammatory diseases [81]. They show *in vitro* antitumor activity against human myelogenous leukemia K562 cells [82] and can be used as HIV-I NCP7 inhibitors [83], or non-vanilloid TRPV1 antagonists [84]. Examples of *S*-alkyl carbodithioate esters (compounds **25-28**) which have potential therapeutic value are shown in Figure 3. These carbodithioate esters are also broadly used as suitable ligands in the area of surface science and nanomaterial chemistry, for the assembly on metal nanoparticles [85, 86]. They are also used as sulfur vulcanization acceptors [87], and radical chain transfer agents in reversible addition fragmentation chain transfer polymerizations in the rubber industry [88-90]. Furthermore, they are also important synthetic intermediates [91-92]. As a result, many methods for the synthesis of these carbodithioate esters have been developed by researchers worldwide [93].

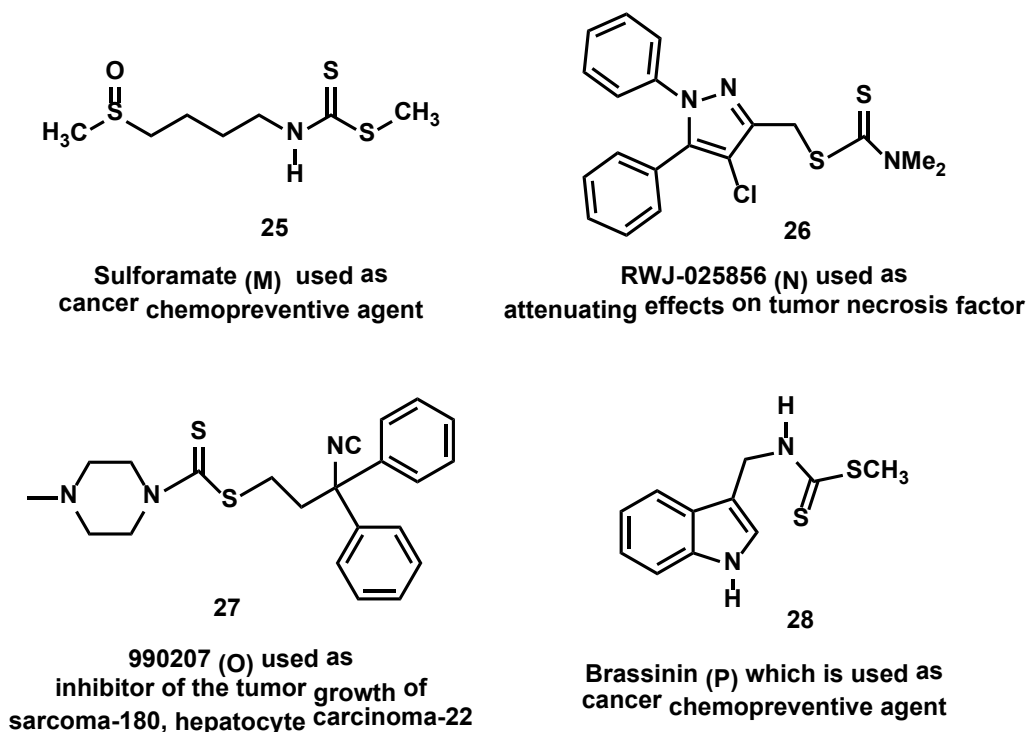


Figure 3. Carbodithioate esters which have the strong potential therapeutic values

4. Conclusions

This review has highlighted the synthesis, reactions and biological importance of α , β -unsaturated carbodithioesters type of compounds and their derivatives. α , β -unsaturated carbodithioesters and their derivatives have myriad applications in biological, pharmaceutical, medicinal and in many other fields. This class of sulfur-containing compounds will continue to be investigated in the future and new applications for these dithioesters will continue to be developed.

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REFERENCES

- [1] a) Kondo, T.; Mitsudo, T-A. *Chem. Rev.* 2000, 100, 3205. b) Norcross, R. D.; Paterson, I. *Chem. Rev.* 1995, 95, 2041. c) Liu, G.; Link, J. T.; Pei, Z.; Reilly, E. B.; Leitza, S.; Nguyen, B.; Marsh, K. C.; Okasinski, G. F.; von Geldern, T. W.; Ormes, M.; Fowler, K.; and Gallatin, M. J. *Med. Chem.* 2000, 43, 4025. d) Sawyer, J. S.; Schmittling, E. A.; Palkowitz, J. A.; Smith, W. J. *J. Org. Chem.* 1998, 63, 6338.
- [2] (a) Metzner, P. *Top. Curr. Chem.* 1999, 204, 127. (b) Boger, D. L. *Tetrahedron.* 1983, 39, 2869. (c) Barluenga, J.; Tomas, M. *Adv. Heterocycl. Chem.* 1993, 57, 1.
- [3] Westmijze, H.; Kleijn, H.; Mijer, J.; Vermeer, P. *Synthesis.* 1979, 6, 432.
- [4] Gosselin, P.; Masson, S.; Thuillier, A. *Tetrahedron Lett.* 1980, 21, 2421.
- [5] (a) Masson, S.; Thuillier, A. *Tetrahedron Lett.* 1982, 23, 4087. (b) Rettberg, N.; Wagner, U.; Hartke, K. *Arch. Pharm. (Weinheim, Ger.)* 1993, 326, 977. (c) Lawson, K. R.; Singleton, A.; Whitham, G. H. *J. Chem. Soc., Perkin Trans. 1*, 1984, 859.
- [6] Hartke, K.; Kunze, O. *Liebigs Ann. Chem.* 1989, 4, 321.
- [7] Hoffman, R.; Hartke, K. *Chem. Ber.* 1980, 113, 919.
- [8] Scheithauer, S.; Mayer, R. 'Topics in Sulphur Chemistry: Thio- and Dithio-carboxylic Acids and Their Derivatives,' ed. A. Senning, Thieme, Stuttgart, 1979, Vol. 4.
- [9] Colonge, J.; Descortes, G. in '1,4-Cycloaddition Reactions,' ed. Hamer, J. Academic Press, New York, 1967, 217; Desimini, G.; Tacconi, G. *Chem. Rev.*, 1975, 75, 651. Weinreb, S. M.; Staib, R. R. *Tetrahedron*, 1982, 38, 3087.
- [10] Meyers, A. I.; Tait, T. A.; Comins, D. L. *Tetrahedron Lett.*, 1978, 4657.
- [11] Westmijze, H.; Kleijn, H.; Meijer, J.; Vermeer, P. *Synthesis*, 1979, 432.
- [12] Hoffmann, R.; Hartke, K. *Chem. Ber.*, 1980, 113, 919.
- [13] Gosselin, P.; Masson, S.; Thuillier, A. *Tetrahedron Lett.*, 1978, 2715; Thuillier, A.; Gosselin, P.; Masson, S. *ibid.* 1980, 21, 2421.
- [14] Colonge, J.; Descortes, G. in '1,4-Cycloaddition Reactions,' ed. Hamer, J. Academic Press, New York, 1967, 217; Desimini, G.; Tacconi, G. *Chem. Rev.*, 1975, 75, 651; Weinreb, S. M.; Staib, R. R. *Tetrahedron*. 1982, 38, 3087.
- [15] Masson, S.; Thuillier, A. *Tetrahedron Lett.*, 1982, 23, 4087.
- [16] Theron, F.; Verny, M.; Vessiere, R. *The Chemistry of the Carbon-Carbon Triple Bond*; Patai, S., Ed.; John-Wiley & Sons: Chichester, 1978, Part 1, Chapter 10.
- [17] Yoshimatsu, M.; Naito, M.; Kawahigashi, M.; Shimizu, H.; Kataoka, T. *J. Org. Chem.* 1995, 60, 4798.
- [18] Hanzawa, Y.; Tabuchi, N.; Taguchi, T. *Tetrahedron Lett.* 1998, 39, 6249.
- [19] Zhongt, P.; Xiongt, Z-X.; and Huang, X. *Synthetic Communications*, 2000, 30, 2793-2800.
- [20] Ruchwald, S. L.; Lamaire, S. J.; Nielsen, R. B.; Watson, B. T. and King, S. M. *Tetrahedron Lett.* 1987, 28, 3895.
- [21] Harpp, D. N.; Friedlander, B. T. and Smith, R. A. *Synthesis*, 1979, 181.
- [22] (a) Des Mazery, R.; Pullez, M.; López, F.; Harutyunyan, S. R.; Minnaard, A. J.; Feringa, B. L. *J. Am. Chem. Soc.* 2005, 127, 9966; (b) Howell, G.P.; Fletcher, S. P.; Geurts, K.; ter Horst, B.; Feringa, B. L. *J. Am. Chem. Soc.* 2006, 128, 14977; (c) van Summeren, R. P.; Moody, D. B.; Feringa, B. L.; Minnaard, A. J. *J. Am. Chem. Soc.* 2006, 128, 4546; (d) ter Horst, B.; Feringa, B. L.; Minnaard, A. J. *Org. Lett.* 2007, 9, 3013.
- [23] Classical methods such as coupling of acids with thiols using DCC/DMAP and transesterification with trimethylsilyl thioethers in presence of $AlCl_3$ occasionally lead to some amount of side product due to 1,4-addition of thiolate to the product (See: Thesis entitled 'Enantioselective copper catalyzed allylic alkylation using Grignard reagents' submitted by van Zijl, AW, University of Groningen).
- [24] van Zijl, A. W.; Minnaard, A. J.; Feringa, B. L. *J. Org. Chem.* 2008, 73, 5651.
- [25] Mohite, A. R.; Mete, T. B.; Bhat, R. G. *Tetrahedron Letters*, 2017, 58, 770-774.
- [26] (a) Yang, W.; Drucekhammer, D. G. *J. Am. Chem. Soc.* 2001, 123, 11004-11009. (b) Wiberg, K. B. *J. Chem. Educ.* 1996, 73, 1089-1095. (c) Cronyn, M. W.; Chang, M. P.; Wall, R. A. *J. Am. Chem. Soc.* 1955, 77, 3031-3034.
- [27] (a) Staunton, J.; Weissman, K. J. *Nat. Prod. Rep.* 2001, 18, 380-416. (b) Stryer, L. *Biochemistry*, 4th ed.; Freeman: New York, 1995. (c) Bruice, T. C.; Benkovic, S. J. *Bioorganic mechanisms*; Benjamin, W. A. New York, 1966, Vol. 1.
- [28] (a) Johnson, J. S.; Evans, D. A. *Acc. Chem. Res.* 2000, 33, 325-335. (b) Fortner, K. C.; Shair, M. D. *J. Am. Chem. Soc.* 2007, 129, 1032-1033. (c) Gennari, C.; Vulpetti, A.; Pain, G. *Tetrahedron* 1997, 53, 5909-5924. (d) Kobayashi, S.; Uchiro, H.; Fujishita, Y.; Shiina, I.; Mukaiyama, T. *J. Am. Chem. Soc.* 1991, 113, 4247-4252. (e) Gennari, C.; Beretta, M. G.; Bernardi, A.; Moro, G.; Scolastico, C.; Todeschini, R. *Tetrahedron*. 1986, 42, 893-909. (f) Evans, D. A.; Nelson, J.

- V.; Vogel, E.; Taber, T. R. J. Am. Chem. Soc. 1981, 103, 3099–3111.
- [29] McGarvey, G. J.; Williams, J. M.; Hiner, R. N.; Matsubara, Y.; Oh, T. J. Am. Chem. Soc. 1986, 108, 4943–4952.
- [30] Fukuyama, T.; Lin, S.-C.; Li, L. J. Am. Chem. Soc. 1990, 112, 7050–7051.
- [31] (a) Wittenberg, R.; Srogl, J.; Egi, M.; Liebeskind, L. S. Org. Lett. 2003, 5, 3033–3035. (b) Liebeskind, L. S.; Srogl, J. J. Am. Chem. Soc. 2000, 122, 11260–11261.
- [32] For a review on thioester chemistry developed in the last 10 years, see: Fujiwara, S.-I.; Kambe, N. Top. Curr. Chem. 2005, 251, 87–140.
- [33] van Zijl, A. W.; Minnaard, A. J.; Feringa, B. L. J. Org. Chem. 2008, 73, 5651–5653.
- [34] Nair, S. K.; Jose, A. M.; Asokan, C. V. Synthesis. 2005, 8, 1261.
- [35] a) Barton, D. H. R.; Chen, M.; Jaszberenyi, J. C.; Rattigan, B.; Tang, D. Tetrahedron Lett. 1994, 35, 6457; (b) Barton, D. H. R.; Jang, D. O.; Jaszberenyi, J. C. Tetrahedron Lett. 1990, 31, 4681; (c) Barton, D. H. R.; Jang, D. O.; Jaszberenyi, J. C. Tetrahedron Lett. 1991, 32, 2569; (d) Barton, D. H. R.; Motherwell, W. B. Pure Appl Chem. 1981, 1.
- [36] Isola, M.; Ciuffarin, E.; Sangramora, L. Synthesis. 1976, 326.
- [37] (a) Degani, I.; Fochi, R. J.; Regondi, V. S. Synthesis. 1981, 149. (b) Baker, R.; Mahony, M.; Sawin, C. J. J. Chem. Soc. Perkin Trans I. 1987, 1623.
- [38] (a) Chugaev, L. Chem. Ber. 1899, 32, 3332. (b) Nace, H. R. Org. React. 1962, 12, 57.
- [39] Barton, D. H. R.; Combie, S. W. J. Chem. Soc. Perkin Trans I. 1975, 1574.
- [40] (a) Tanaka, K.; Yamagishi, N.; Tanikaga, R.; Kaji, A. Chem. Soc. Japan. 1979, 52, 3619. (b) Degani, I.; Fochi, R.; Regondi, V. S. Synthesis. 1979, 178.
- [41] Okawata, M.; Nakai, T.; Otsuji, Y.; Imoto, E. J. Org. Chem. 1965, 30, 2025. (b) Barton, D. H. R. Tetrahedron. 1992, 48, 2529. (c) Zard, S. Z. Angew Chem. Int. Ed. 1997, 36, 672.
- [42] Nieuwenhuizen, P. J.; Ethers, A. W.; Haasnoot, J. G.; Janse, S. R.; Reedijk, J.; Baerends, E. J. Am. Chem. Soc. 1999, 121, 163.
- [43] Zhang, D.; Chen, J.; Liang, Y.; Zhou, H. Synth. Commun. 2005, 35, 521.
- [44] Curren, D. P. Synthesis. 1988, 417, 489.
- [45] (a) Ferrier, R. J.; Vethavisar, V. Chem Commun. 1970, 1385. (b) Baldwin, J. E.; Holfe, G. A. J. Am. Chem. Soc. 1971, 93. (c) Nakai, T.; Ari- Izumi, A. Tetrahedron Lett. 1976, 2335.
- [46] (a) Meurling, P.; Sjöberg, K.; Sjöberg, B.; Acta Chem Scand. 1972, 26, 279. (b) Mori, T. M.; Taguchi, T. Synthesis. 1975, 469. (c) Degani, I.; Foch, R.; Santi, M. Synthesis. 1977, 873.
- [47] Chaturvedi, D.; Ray, S. J. Sulfur Chem. 2006, 27, 265.
- [48] Chaturvedi, D.; Ray, S. Monatsh Chem. 2006, 137, 1219.
- [49] (a) Ugi, I.; Domling, A.; Hori, W. Endeavour. 1994, 18, 115. (b) Armstrong, R. M.; Combs, A. P.; Tempest, P. A.; Brown, S. D.; Keating, T. A. Acc. Chem. Res. 1996, 29, 123.
- [50] Bienayme, H.; Hume, C.; Oddon, G.; Schmidt, P. Chem Eur J. 2000, 6, 3321.
- [51] (a) Nielsen, T. E.; Schrieber, S. L. Angew Chem Int Ed. 2008, 47, 48. (b) Domling, A.; Ugi, I. Angew Chem. Int. Ed. 2000, 39, 48.
- [52] Ganem B. Acc. Chem. Res. 2009, 42, 463.
- [53] Patra, G. C.; Pal, S.; Bhunia, S. C.; Hazra, N. K.; Pal, S. C. Ind. J. Chem. 2016, 55B, 471–477.
- [54] Biswas, K.; Ghosh, S.; Ghosh, P.; Basu, B. Cyclic ammonium salts of dithiocarbamic acid: stable alternative reagents for the synthesis of S-alkyl carbodithioates from organyl thiocyanates in water, Journal of Sulfur Chemistry, 2016, 37(4), 361–376.
- [55] Guzman, M. L.; Rossi, R. M.; Karnischky, L. *et al.* The sesquiterpene lactone parthenolide induces apoptosis of human acute myelogenous leukemia stem and progenitor cells. Blood 2005, 105, 4163–4169.
- [56] Siveen, K. S.; Uddin, S.; Mohammad, R. M. Targeting acute myeloid leukemia stem cell signaling by natural products. Mol Cancer, 2017, 16, 13.
- [57] Sarkozy, C.; Gardin, C.; Gachard, N.; *et al.* Outcome of older patients with acute myeloid leukemia in first relapse. Am J Hematol, 2013, 88, 758–764.
- [58] Lapidot, T.; Sirard, C.; Vormoor, J. *et al.* A cell initiating human acute myeloid leukaemia after transplantation into SCID mice. Nature, 1994, 367, 645–648.
- [59] Sarry, J.-E.; Murphy, K.; Perry, R. *et al.* Human acute myelogenous leukemia stem cells are rare and heterogeneous when assayed in NOD/SCID/IL2R gamma c-deficient mice. J Clin Invest 2011, 121, 384–395.
- [60] Stiehl, T.; Baran, N.; Ho, A. D.; Marciniak-Czochra, A. Cell division patterns in acute myeloid leukemia stem-like cells determine clinical course: a model to predict patient survival. Cancer Res. 2015, 75, 940–949.
- [61] Ho, T.-C.; LaMere, M.; Stevens, B. M. *et al.* Evolution of acute myelogenous leukemia stem cell properties following treatment and progression. Blood, 2016, 128, 1671–1678.
- [62] Shlush, L. I.; Mitchell, A.; Heisler, L. *et al.* Tracing the origins of relapse in acute myeloid leukaemia to stem cells. Nature, 2017, 547, 104–108.
- [63] Thomas, D.; Majeti, R. Biology and relevance of human acute myeloid leukemia stem cells. Blood, 2017, 129, 1577–1585.
- [64] Eppert, K.; Takenaka, K.; Lechman, E. R. *et al.* Stem cell gene expression programs influence clinical outcome in human leukemia. Nat Med. 2011, 17, 1086–1091.
- [65] Chan, W. I.; Huntly, B. J. Leukemia stem cells in acute myeloid leukemia. Semin Oncol. 2008, 35, 326–335.
- [66] Jin, L.; Hope, K. J.; Zhai, Q. *et al.* Targeting of CD44 eradicates human acute myeloid leukemia stem cells. Nat Med. 2006, 12, 1167–1174.
- [67] Dinga, Y.; Yanga, Z.; Gea, W.; Kuanga, B.; Xuc, J.; Yanga, J.; Chena, Y.; Zhang, Q. Journal of Enzyme Inhibition and Medicinal Chemistry, 2018, 33(1), 1376–1391.

- [68] Ware, G. W.; Whitcare, D. M. The pesticide book, 6th ed. Willoughby: Meister Pro Information Resources; 2004.
- [69] Caldas, E. D.; Conceição, M. H.; Miranda, M. C. C.; de Souza, L.C.K.R.; Lima, J.F. Determination of dithiocarbamate fungicide residues in food by a spectrophotometric method using a vertical disulfide reaction system. *J Agric Food Chem.* 2001, 49, 4521–4525.
- [70] Rafin, C.; Veignie, E.; Sancholle, M. Synthesis and antifungal activity of novel bisdithiocarbamate derivatives of carbohydrates against *fusarium oxysporum* f. sp. *lini*. *J Agric Food Chem.* 2000, 48, 5283–5287.
- [71] Perz, R.C.; van Lishaut, H.; Schwack, W. CS₂ blinds in brassica crops: false positive results in the dithiocarbamate residue analysis by the acid digestion method. *J Agric Food Chem.* 2000, 48, 792–796.
- [72] Husain, A.; Nami, S.A.A.; Singh, S.P.; Oves, M.; Siddiqi, K.S. Anagostic interactions, revisiting the crystal structure of nickel dithiocarbamate complex and its antibacterial and antifungal studies. *Polyhedron.* 2011, 30, 33–40.
- [73] Shaheen, F.; Badshah, A.; Gielen, M. *et al.* Synthesis, characterization, antibacterial and cytotoxic activity of new palladium (II) complexes with dithiocarbamate ligands: X-ray structure of bis(dibenzyl-1-S:S'-dithiocarbamate) Pd (II). *J Organomet. Chem.* 2007, 692, 3019–3026.
- [74] Manav, N.; Mishra, A.K.; Kaushik, N.K. In vitro antitumor and antibacterial studies of some Pt(IV) dithiocarbamate complexes. *Spectrochim Acta Part A.* 2006, 65, 32–35.
- [75] Yurttas, L.; Ozkay, Y.; Demirci, F. *et al.* Synthesis, anticandidal activity, and cytotoxicity of some thiazole derivatives with dithiocarbamate side chains. *Turk J Chem.* 2014, 38, 815–824.
- [76] Alagarsamy, V.; Narendhar, B.; Sulthana, M.T.; Solomon, V.R. Design and synthesis of 3-(4-chlorophenyl)-2-(2-(4-substituted)-2-oxoethylthio)quinazolin-4(3H)-one as antihistamine agents. *Med Chem Res.* 2014, 23, 4692–4699.
- [77] Ghorbani-Vaghei, R.; Amiri, M.; Veisi, H. *et al.* A new and facile protocol for the synthesis of dithiocarbamate-linked 3,4-dihydro-2H-pyran using *N*-halo catalysts under mild conditions reaction. *Bull Korean Chem Soc.* 2012, 33, 4047–4051.
- [78] Cui, J.-L.; Ge, Z.-M.; Cheng, T.-M.; Li, R.-T. An efficient one-pot synthesis of 2-hydroxyalkyl dithiocarbamates. *Synth Commun.* 2003, 33, 1969–1976.
- [79] Scozzafava, A.; Mastrolorenzo, A.; Supuran, C.T. Arylsulfonyl-*N,N*-diethyl-dithiocarbamates: a novel class of antitumor agents. *Bioorg Med Chem Lett.* 2000, 10, 1887–1891.
- [80] Hawthorne, M.; Mehta, R.G.; Moon, R.C.; Pezzuto, J.M. Cancer chemopreventive potential of sulforamate, a novel analogue of sulforaphane that induces phase 2 drug-metabolizing enzymes. *Cancer Res.* 1997, 57, 272–278.
- [81] Medford, R.M.; Saxena, U.; Hoong, L.K.; Somers, P.K. *N*-substituted dithiocarbamates for the treatment of biological disorders. United States patent US 6,747,061B2. 2004 Jun 8.
- [82] Cao, S.-L.; Feng, Y.-P.; Jiang, Y.-Y. *et al.* Synthesis and in vitro antitumor activity of 4(3H)-quinazolinone derivatives with dithiocarbamate side chains. *Bioorg Med Chem Lett.* 2005, 15, 1915–1917.
- [83] Goel, A.; Majur, S.J.; Fattah, R.J. *et al.* Benzamide-based thiolcarbamates: a new class of HIV-1 NCp7 inhibitors. *Bioorg Med Chem Lett.* 2002, 12, 767–770.
- [84] Suh, Y.-G.; Lee, Y.-S.; Min, K.-H. *et al.* Novel potent antagonists of transient receptor potential channel, vanilloid subfamily member 1: structure-activity relationship of 1,3-diarylalkyl thioureas possessing new vanilloid equivalents. *J Med Chem.* 2005, 48, 5823–5836.
- [85] Guerrini, L.; Garcia-Ramos, J.V.; Domingo, C.; Sanchez-Cortes, S. Sensing polycyclic aromatic hydrocarbons with dithiocarbamate-functionalized Ag nanoparticles by surface-enhanced Raman scattering. *Anal Chem.* 2009, 81, 953–960.
- [86] Zhao, Y.; Pérez-Segarra, W.; Shi, Q.; Wei, A. Dithiocarbamate assembly on gold. *J Am Chem Soc.* 2005, 127, 7328–7329.
- [87] Nieuwenhuizen, P.J.; Ehlers, A.W.; Hassnot, J.G. *et al.* The mechanism of zinc(II)-dithiocarbamate accelerated vulcanization uncovered; theoretical and experimental evidence. *J Am Chem Soc.* 1999, 121, 163–168.
- [88] Bathfield, M.; D'Agosto, F.; Spitz, R.; Charreyre, M.T.; Delair, T. Versatile precursors of functional RAFT agents. Application to the synthesis of bio-related end-functionalized polymers. *J Am Chem Soc.* 2006, 128, 2546–2547.
- [89] Lai, J.T.; Shea, R.J. Controlled radical polymerization by carboxyl- and hydroxyl-terminated dithiocarbamates and xanthates. *Polym Sci Part A: Polym Chem.* 2006, 44, 4298–4316.
- [90] Dureault, A.; Gnanou, Y.; Taton, D.; Destarac, M.; Leising, F. Reaction of cyclic tetrathio phosphates with carboxylic acids as a means to generate dithioesters and control radical polymerization by RAFT. *Angew Chem Int Ed.* 2003, 42, 2869–2872.
- [91] Boas, U.; Gertz, H.; Christensen, J.B.; Heegaard, P.M.H. Facile synthesis of aliphatic isothiocyanates and thioureas on solid phase using peptide coupling reagents. *Tetrahedron Lett.* 2004, 45, 269–272.
- [92] Mukerjee, A.K.; Ashare, R. Isothiocyanates in the chemistry of heterocycles. *Chem Rev.* 1991, 91, 1–24.
- [93] Aly, A.A.; Brown, A.B.; Bedair, T.M.I.; Ishak, E.A. Dithiocarbamate salts: biological activity, preparation, and utility in organic synthesis. *J Sulfur Chem.* 2012, 33, 605–617.