

Accumulation of Selenium in Fathead Minnows (*Pimephales Promelas*)

Shahrokh Ghaffari*, Shadi Abu-Baker*, Erin Glynn

Department of Chemistry and Biochemistry, Ohio University, Zanesville, Ohio, USA

Abstract The purpose of conducting this undergraduate research project is to test the amount of selenium absorbed by fathead minnows when selenium is added to the water supply. Selenium is a nonmetallic element that is found in a variety of chemical forms both in marine and freshwaters (3). In this study there were two groups of fathead minnows studied. One was the control and the other contained 20 ppm of selenium in selenite form. The fathead minnows were placed in a 25-degree Celsius temperature-controlled environment and monitored for five days. The minnows that expired during this time were frozen in plastic bags. After the five days the remaining fish were terminated by exposure. The minnows were converted to ashes in a muffle furnace and prepared following the Yankton Biological Research Field Stations SOP F6.3.13. The samples were then analyzed by an atomic absorption spectrophotometer 200-A. The results showed a substantial difference between the control and experimental groups, thus rendering our hypothesis as accurate.

Keywords Selenium absorption, Marine and freshwaters ecosystem, Fathead minnows' ashes, Undergraduate research, and atomic absorption spectrophotometer

1. Introduction

The purpose of conducting this undergraduate research study on the accumulation of selenium in fathead minnows were to discover the amount of selenium absorbed and thus accumulated in the food chain. Selenium is found in both marine and freshwaters [1]. Selenium can exist in multiple oxidation states such as selenate, Se (VI); selenite, Se (IV); elemental selenium, Se (0); and selenide, Se (-II). It can also be found in multiple chemical forms within an oxidation state, such as organic and inorganic selenide [2]. Selenium is necessary in small amounts, but it can have adverse and be potentially fatal if consumed in large quantities. Waterborne selenium with concentrations of 2 mg/L or greater should be considered hazardous to the health and long-term survival of fish and wildlife populations due to the high potential for food-chain bioaccumulation, dietary toxicity, and reproductive effects [3]. It is important to know how much bioaccumulation will take place in a particular species before estimating the dietary toxicity and reproductive effects. According to Lemly, field studies show that benthic invertebrates and certain forage fishes, such as fathead minnows, can accumulate 20 to 370 mg/g selenium and still maintain

stable, reproducing populations. To analyze the amount of selenium found in the samples there are several recommended methods. They include hydride generation atomic absorption spectrometry, neutron activation analysis, fluorometry, and gas chromatography [4].

2. Methods and Results

2.1. Methods

An This study was conducted using two 20-gallon fish aquariums filled with dechlorinated water. One tank was used as a control and the other as the experimental. The experimental tank had 20 ppm of sodium selenite, 99%, (Na_2SeO_3) added. Ten fathead minnows (*Pimephales promelas*, Figure 1) were placed in each tank. We picked this species as it was available used in literature before [4]. The water temperature was regulated to 20 degrees Celsius. The minnows were fed a few flakes of fish food every day. They were alive for five days. At the end of this time period the remaining fish were terminated by exposure. The fathead minnows were vacu-washed, freeze-dried and the percent moisture of the samples is presented in Table 1. All the minnows were then ashed following the "Yankton Biological Research Field Stations SOP F6.3.13 procedure". After the digestion of the minnows was completed, they were analyzed by an atomic absorption spectrophotometer 200-A. To produce a Calibration Curve four standards solution of selenium were prepared consisting of 2.5, 5, 10, and 25 ppm. The calculations required for the addition of selenium to

* Corresponding author:
ghaffari@ohio.edu (Shahrokh Ghaffari)
abu@ohio.edu (Shadi Abu-Baker)

Received: Jun. 27, 2021; Accepted: Jul. 21, 2021; Published: Jul. 30, 2021
Published online at <http://journal.sapub.org/ajee>

conduct this study are as follows:

Amount of Se added to experimental tank:

$3.4 \text{ L} \times 40 \text{ mg/L} = 136 \text{ mg Se}$. Then to calculate Se in grams and ppm, $(0.136 \text{ g Se} / 46.6\%) / 100 = 0.292 \text{ g Se}$ 20 ppm = $0.292 \text{ g Se} / 2 = 0.146 \text{ g Se}$.

2.2. Results

The results indicate that the fathead minnows in the experimental tank did have a significant absorption of selenium in comparison with the control tank. The following table shows the concentrations and absorptions of the fish samples.

Table 1. Data presented here are averages of 10 measurements, Average absorbance and Se concentration in ppm of both treated and control samples in 100 ml prepared sample as well as in wet fish

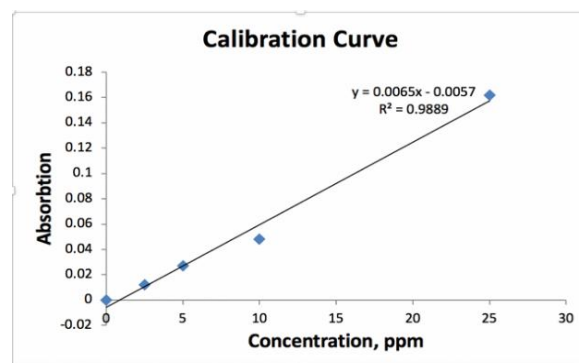
Measurement/amount	Value
Average masses of sample and calculated water content	
Weight of samples (g)	4.2881 ± 1.2879
Weight of dry sample (g)	$0.8330 \pm 0.3559 \text{ (g)}$
Weight of water (g)	3.4556 ± 0.4904
Percent water (w/w) %	81.131 ± 4.347
Average absorbance and Se concentration in 100 mL samples calculated in ppm	
Absorbance of control samples	0.091 ± 0.018
Amount of Se in 100ml control samples (ppm)	10.98 ± 0.961
Absorbance of sample with added Se	0.066 ± 0.007
Amount of Se in 100ml sample with added Se (ppm)	18.22 ± 2.01
Average Concentration of Se in wet fish in ppm	
Amount of Se in wet fish control samples (ppm)	0.471 ± 0.042
Amount of Se in wet fish sample with added	0.781 ± 0.086

The concentration of the samples with selenium are almost double the concentrations of the samples without selenium. The absorbance of the samples with exposure to selenium was about 0.1 ppm were the samples without exposure to selenium were about 0.06 ppm. This is about twice as much as control samples.

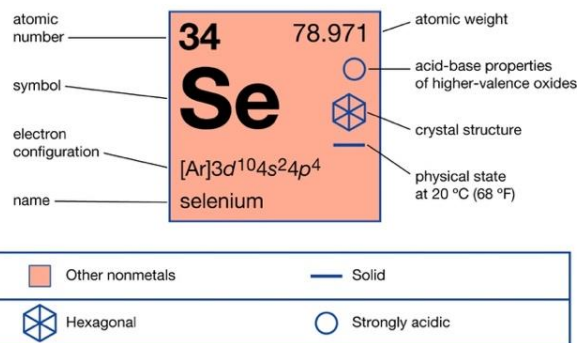
3. Discussion and Conclusions

This experiment was a success in the fact that the hypothesis of increased selenium absorption in fathead minnows was accurate. According to Sanders, selenium is incorporated into the amino acids, and its transfer and depuration in the microbial food web may be analogous to those of nitrogen in some ways [1]. Thus, selenium is concentrated into the tissues of the fathead minnows. This value was measured by the atomic absorption spectrophotometer and indicates the increased amount of selenium found in the experimental tank compared to the control tank. "Most selenium accumulation occurs at the base of the food web and selenium concentration in predators

generally reflects the concentration in their prey" [1].



Selenium



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Figure 1. Calibration curve of Se concentration in ppm vs. the absorbance, the image of **Fathead Minnow (*Pimephales promelas*)** and were obtained from online resource [6,7]

The fathead minnows used in this study would be considered to be close to the base of the food web and therefore they would bioaccumulate selenium at a higher concentration than a larger predator. The chemistry of the water ecosystem also reflects the habitat of fish. The ecosystem used in this experiment was very narrow and selective. "Scientists now understand that water chemistry and biochemical transformations of selenium influence its accumulation in the tissues of aquatic life-forms-and therefore determine its impact on the ecology of lakes and streams" [2]. By understanding the cycling of various chemicals in the ecosystem it is easier to predict the impact an increased amount of a certain chemical might have. For example, the increased amount of selenium found in the experimental tank had an adverse effect on the minnows by causing death in a short amount of time. This was not a gradual increase, but a more dramatic increase and thus detrimental. This research was conducted to discover and

evaluate what an increased amount of selenium in the water supply would accumulate in the body tissue of fathead minnows. It has been reported that acute toxicity has been seen in regions with high selenium content in food, and the main cellular toxopathology is not currently well understood. Some studies suggested mechanisms including the formation of superoxide and hydroxyl anions, together with hydrogen peroxide [8]. This study could expand to include toxicity studies of sodium selenite and sodium selenate acute toxicity other species as reported [9]. We could also study the effect of different concentrations of selenium on different species of fish as reported in the literature [10].

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